

## Lychnophoric acid from *Lychnophora pinaster*: a complete and unequivocal assignment by NMR spectroscopy.

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**Abstract:** The investigation of the hexane extract from aerial parts of *Lychnophora pinaster* provided, besides others substances, the *E*-isomer of lychnophoric acid, a sesquiterpene derivative previously isolated from *L. affinis*.

**Keywords:** *Lychnophora pinaster*; Asteraceae; lychnophoric acid.

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### Introduction

Plant species of the genus *Lychnophora* (*Asteraceae*) are known as “candeia”, “arnica” and “arnica da serra” and are used in folk medicine as anti-flogistic, anti-rheumatic, and analgesic [1]. Typical constituents of *Lychnophora* species are sesquiterpene lactones [2] of which 15-deoxygoyazensolide was shown to be active against *Trypanosoma cruzi*, the etiological agent of Chagas’ disease (American trypanosomiasis) [3]. Prompted by this observation we have carried out a screening of *Asteraceae* plant species in the search of new trypanocidal agents [4] and we have investigated three active *Lychnophora* species, one of them being *L. pinaster* Mart. Bioguided fractionation of the hexane and dichloromethane extracts of the aerial parts of this plant [5] led to the isolation of lychnophoric acid (*I*), previously isolated from *L. affinis*, that was assayed *in vitro* against bloodstream forms of *T. cruzi* and presented 50% growth inhibition in the dose of 12,0mg/mL [6].

### Experimental

#### General

Melting point was determined on a Mettler FP5 apparatus;  $[\alpha]_D^{25}$  was measured at 25 °C on a Bellincham & Stanley Ltd P-20 polarimeter. IR spectrum was obtained on a Shimadzu/IR-408 spectrometer. EIMS was obtained on a Kratos MS 80 RFA spectrometer. <sup>1</sup>H and <sup>13</sup>C NMR spectra and contour plots were acquired on a Bruker AVANCE DRX400 instrument operating at 400 MHz for <sup>1</sup>H and 100 MHz for <sup>13</sup>C. HPLC analysis was performed with a Shimadzu CR-8, UV detector. CG analysis was performed with a HP5890 gas chromatograph, FID detector, and a VDC3390A integrator.

#### Plant material

The aerial parts of *Lychnophora pinaster* Mart. were collected at Serra da Moeda, State of Minas Gerais, Brazil, in March 1992. A voucher specimen has been deposited in the Herbarium of

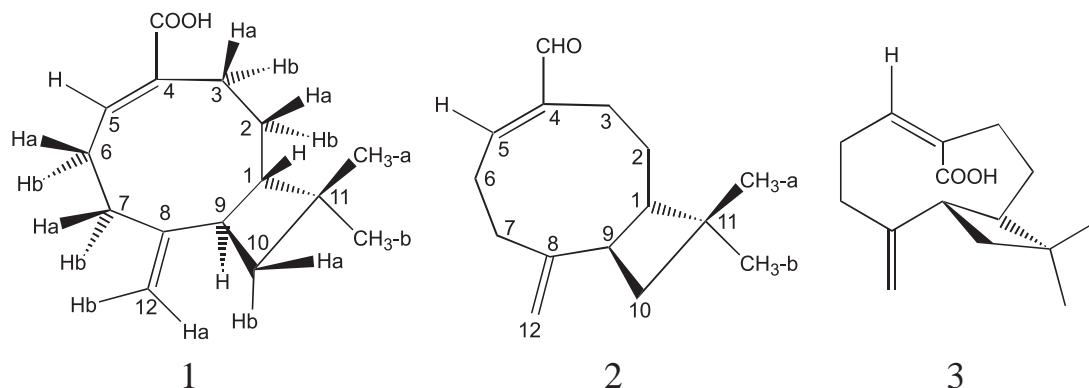
the Instituto de Ciências Biológicas, UFMG, Belo Horizonte, Minas Gerais (BHCB-UFMG 19520).

### Extraction procedures

The dried aerial parts (2.0 Kg) were powdered and successively extracted with n-hexane and dichloromethane. The solvents were removed under vacuum, below 40 °C, to give 114.0 g of n-hexane and 12.0 g of dichloromethane extracts.

The crude extracts were chromatographed first by CC (Silica gel 60, hexane-CH<sub>2</sub>Cl<sub>2</sub>-AcOEt-MeOH gradient). The n-hexane extract (114.0 g) furnished a homologue series of saturated hydrocarbons (C<sub>22</sub>-C<sub>32</sub>) [7], lupeol, a- and b-amyryn, friedelin and fat acid esters detected by GC, in comparison with authentic samples. The CH<sub>2</sub>Cl<sub>2</sub> fr. was chromatographed over florisil column. Fraction 1 (petrol), after washing with Et<sub>2</sub>O-MeOH (1:1), filtration and solvent evaporation, afforded a yellow gum, which was partitioned between hexane and MeOH-H<sub>2</sub>O (9:1). The MeOH-H<sub>2</sub>O fr., after 4 days at 4 °C, afforded *I*. CC of the CH<sub>2</sub>Cl<sub>2</sub> extract (12.0 g) afforded a homologue series of saturated hydrocarbons (C<sub>25</sub>-C<sub>32</sub>) [7] detected by GC, as well quercetin and 15-deoxygoyazensolide, detected by HPLC, using authentic samples as standard.

*E*-Lychnophoric acid (*1*): Bicyclo [7.2.0] undec-4-en-4-carboxylic acid-11,11-dimethyl-8-methylen-[1*R*-(1*R*\*,4*E* ,9*S*\*)]. Amorphous solid, mp 118-9 °C (Et<sub>2</sub>O), [α]<sub>D</sub><sup>20</sup> = -24° (CHCl<sub>3</sub>; c=0,054). IR n<sub>max</sub> cm<sup>-1</sup> 3050-2400, 2900, 1680 (C=CCO<sub>2</sub>H); 1640 (C=CH<sub>2</sub>), 890. EIMS *m/z* (rel. int.): 254 [M<sup>+</sup>] (15) (C<sub>15</sub>H<sub>22</sub>O<sub>2</sub>), 219 [M-Me] (25), 69 (C<sub>5</sub>H<sub>9</sub><sup>+</sup>) 100. <sup>1</sup>H NMR and <sup>13</sup>C NMR (see Table 1).



*Quercetin*: R<sub>t</sub> = 17.16 min. HPLC conditions: LiChroCART 125-4 RP-18 column; MeCN/H<sub>2</sub>O gradient, 15 to 45%, 30 min.

*15-deoxygoyazensolide*: R<sub>t</sub> = 9.19 min. HPLC conditions: LiChroCART 125-4 RP-18 column; Hexane-CH<sub>2</sub>Cl<sub>2</sub> (3:7) isocratic, 0.5 mL/min.

### Results and discussion

The hexane extract from the dried aerial parts of *L. pinaster* was column chromatographed over silica gel affording mixtures of homologue hydrocarbons [7], triterpenes (lupeol, a- and b-amyryn, friedelin), fat acids (identified by GLC of their methyl esters), and a caryophyllene derivative, lychnophoric acid (*1*). The dichloromethane extract afforded a mixture of homologue hydrocarbons. Quercetin and 15-deoxygoyazensolide were detected by HPLC in comparison with authentic samples.

The IR spectrum of compound *1* showed absorption bands due to conjugated carboxylic function group (3600-2400, 1680 cm<sup>-1</sup>), carbon-carbon double bonds (1640, 1470, 890 cm<sup>-1</sup>), and *gem*-dimethyl groups (1370 cm<sup>-1</sup>). Its <sup>1</sup>H NMR spectrum (Table 1) exhibited characteristic signals indicating the presence of a terminal olefinic methylene group (δ 4.87 and δ 4.81) and another olefinic hydrogen in an a,b-unsaturated carboxylic group (δ 7.00). Two 3H singlets at δ 0.96 and δ 1.00 confirmed a *gem*-dimethyl group. EIMS indicated a [M]<sup>+</sup> of *m/z* 254, which in conjunction with <sup>1</sup>H and <sup>13</sup>C NMR data allowed the assignment of the molecular formula C<sub>15</sub>H<sub>22</sub>O<sub>2</sub> to (*1*). These data are very similar to those reported for lychnophoric acid (*3*) [8,9].

**Table 1:** NMR\* data ( $\delta$ ) from lychnophic acid (1), Isocaryophyllen-13-al (2) and lychnophoric acid (3).

	<b>1</b>	<b>2</b> [10]	<b>3</b> [8]	<b>1</b>	<b>2</b> [10]	<b>3</b> [9]
<b>1</b>	1.81 (ddd)	1.70	1.65 (m)	51.90	52.38	52.10
<b>2</b>	a:1.48 (dddd); b:1.67 (ddt)	1.44; 1.65	1.45 (m)	27.30	27.10	27.40
<b>3</b>	2.33 (m); 2.43 (m)	2.39; 2.28	2.25 (m)	23.70	21.89	23.70
<b>4</b>	-		-	132.00	144.20	132.20
<b>5</b>	7.00 (t <sup>†</sup> )	6.54	6.22 (m)	144.70	154.46	144.70
<b>6</b>	2.30 (m); 2.41 (m)	2.65; 2.36	2.25 (m)	33.90	28.81	34.00
<b>7</b>	2.41 (m); 2.50 (m)	2.50; 2.36	2.33 (m)	28.50	34.19	28.50
<b>8</b>	-	-	-	154.50	153.94	154.40
<b>9</b>	2.50(q <sup>‡</sup> )	2.43	2.65 (m)	40.10	40.97	40.20
<b>10</b>	a:1.73(ddd); b:1.57(dd)	a: 1.57; b:1.71	1.65 (m)	40.20	40.04	40.30
<b>11</b>	-	-	-	33.20	33.48	33.30
<b>12</b>	a: 4.87(dd); b: 4.81(m)	a:4.86; b:4.82	5.03 (d); 4.88 (d)	111.40	111.64	111.50
<b>Me</b>	a:1.00(s); b:0.96(s)	a:0.94; b:0.98	1.02 (s); 1.00 (s)	a:29.90 ; b:22.80	a:30.04 ; b:22.73	a:30.00; b:22.90
<b>CO</b>	-	-	-	173.00	195.48	173.80

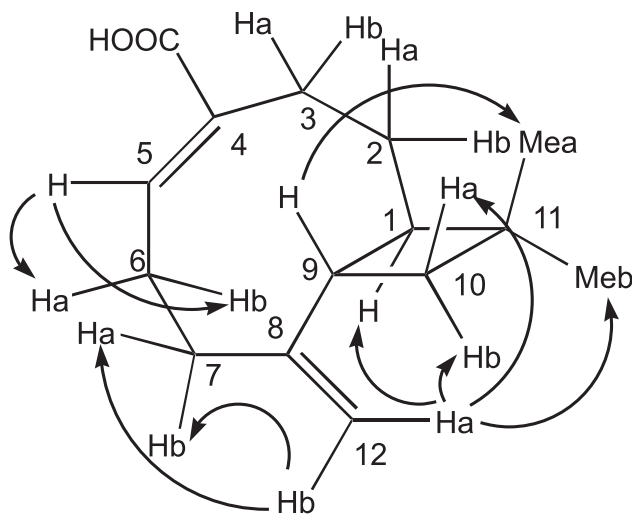
1: 400MHz (<sup>1</sup>H); 100MHz(<sup>13</sup>C); 2: 500MHz (<sup>1</sup>H); 125MHz(<sup>13</sup>C); 3: 200MHz (<sup>1</sup>H); 50MHz (<sup>13</sup>C);\* TMS as internal standard; <sup>†</sup> apparent triplet; <sup>‡</sup> apparent quartet;

Coupling Constants (Hz): In parentheses are the analogous values for 2 and 3, respectively.  $J_{1,2a} = 12.0$ ;  $J_{1,2b} = 3.8$ ;  $J_{1,9} = 9.2$ ;  $J_{1,10a} = 0.7$ ;  $J_{2a,2b} = 13.9$ ;  $J_{2a,3a} = 7.6$ ;  $J_{2a,3b} = 12.0$ ;  $J_{2b,3a} = 9.1$ ;  $J_{2b,3b} = 3.8$ ;  $J_{5,6a} = 7.8$ ;  $J_{5,6b} = 9.3$ ;  $J_{9,10a} = 9.4$ ;  $J_{9,10b} = 9.4$ ;  $J_{10a,10b} = 10.9$ ;  $J_{12a,7a}$  and  $J_{12a,7b} = 1.6$  or 0.8.

However, divergences between **1** and **3** were observed for the  $^1\text{H}$  NMR data: the signal of H-5 is shifted to a higher value of  $\delta$  7.00 in the former, in comparison to that one originally described for lychnophoric acid ( $\delta$  6.22) [8]. This fact can be explained by the change in the configuration of the double bond from *Z*-configuration in **3** to *E*-configuration in **1**, where the closer carbonyl group can contribute with its stereoelectronic deshielding effect. Besides the difference in chemical shifts, a difference in the multiplicity of the H-5 signal in the two compounds is also observed. In the *Z*-isomer (**3**), this signal is described as a multiplet due to coupling with the two adjacent H-6 and to a long-range coupling with two allylic H-3 [8]. The *E*-isomer (**1**)  $^1\text{H}$  NMR spectrum shows an apparent triplet ( $\delta$  7.00,  $J=7.8$  Hz and  $J=9.3$  Hz) for H-5 due to imperfect superposition of the two inner signals of the theoretical double doublet, and the long range

coupling with the two H-3 is not observed.

Despite the use of the Gaussian multiplication with Trafficante function altogether in the normal fid, we could not achieve enough improvement of resolution to picture the H-5 theoretical double doublet. Likewise for the compound **2**, the signal of H-9 appears as a quartet. All chemical shifts were supported by one and two-dimensional NMR techniques like NOEDIFF, COSY and NOESY. In particular the HMQC experiment was very important to the assignments of the chemical shifts inside the complex envelopes. For example, a strong nOe were observed for the protons H-12a ( $\delta$  4.87) with H-10b ( $\delta$  1.57), H-10a ( $\delta$  1.73), H-9 ( $\delta$  2.50) and Me-b group ( $\delta$  1.00) and between the protons H-9 ( $\delta$  2.50) and Me-a ( $\delta$  0.96) as well for the H-12b ( $\delta$  4.81) with H-7a,b spin system. The nOe were also observed for H-5 ( $\delta$  7.00) and H-6a,b system. The nOe results are summarized in the figure 1.



**Figure 1.** nOe assignments for lychnophoric acid (**1**) by NOESY experiment (ns 16, ds 4, d8 0.5 sec, TD 2K)

## Conclusions

The  $^{13}\text{C}$  NMR data for compound **1** are very close to those reported for compound **3** [9] (TABLE 1). The authors [9] did not report the  $^1\text{H}$  NMR data.

These data led us to consider (**1**), is in fact the *E*-isomer of lychnophoric acid, originally described as the *Z*-isomer (**3**) in reference 8. Based on the reported  $^{13}\text{C}$  NMR data (Table 1) the

compound reported also represents the *E*-isomer (**1**) instead of the *Z*-isomer (**3**), as previously proposed [9].

The spectral data and nOe results of **1** are in good accord with data reported for aldehyde **2** [10].

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D. Silveira, J. D. de Souza Filho, A. B. de Oliveira, D. S. Raslan. Atribuição completa e inequívoca dos sinais de deslocamento químico dos átomos de carbono e hidrogênio do ácido licnofórico extraído de *Lychnophora pinaster*.

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**Resumo:** O estudo químico das partes aéreas do extrato hexânico de *Lychnophora pinaster* forneceu, além de outras substâncias, o isômero *E* do ácido licnofórico, um sesquiterpeno anteriormente isolado de *L. affinis*.

*Palavras-chave:* *Lychnophora pinaster*; Asteraceae; ácido licnofórico.

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